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Name _____ Date _____ Period ____ Chapter 17 Thermochemistry Study Guide 17.1 – 17.2 Thermochemical Equations 1. Make the following conversions: a. 444 cal to joules = 1.86 x 103 J b. 1.8 kJ to joules = 1.8 x 103 J c. 0.45 kJ to calories = 1.1 x 102 cal 2. Classify each of these processes as endothermic or exothermic:

a. condensing steam ...

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Chapter 17 Thermochemistry Study Guide. Chapters 15 & 16 Thermochemistry Study Guide. You must show all work and setup for this to count as extra credit on your test (+3 points) 15.1 – 15.2 Heat, Calorimetry, and Enthalpy. Make the following conversions: 444 cal to joules. 1.8 kJ to joules. 0.45 kJ to calories.

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distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)

SECTION 17.1 THE FLOW OF ENERGY-HEAT AND WORK(pages505-510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy Transformations(page505) 1.

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Chemistry Thermochemical Equations Study Guide Answers

Chapter 16: Thermochemistry I Thermochemistry A. “Thermo” refers to heat B. Thermochemistry: the study of the transfers of energy as heat in chemical reactions and physical changes. Energy is either gained or lost II Temperature and heat are related, but not identical A. Heat 1. Heat is a form of energy 2.